## 1ST SEM /COMMON TO ALL BRANCHES/ 2020(W) NEW

## TH2-B ENGINEERING CHEMISTRY

Full Marks: 80

Time- 3 Hrs

Answer any five Questions including Q No.1& 2 Figures in the right hand margin indicates marks

Answer All questions.

2 x 10

- 1. %a. What do you mean by 'penultimate shell'? what is the maximum capacity of this shell to hold electrons?
  - 6. What are basic salts? Give an example of it.
  - What happens in the reduction step of inetallurgical operation?
  - d. Give the bond-line formula of 3-Methylpent-2-ene.
  - 9. Which chemical substances are responsible for the temporary hardness of water?
  - f. Write down the electronic configurations of Cu and Fe<sup>11</sup> ion.
  - What is the major component of CNG? Give one use of CNG.
  - h./ Name the monomers of Bakelite.
  - Define insecticides. Give example of an insecticide.
    - What are lubricants? Give an example of solid lubricant.
- 2. Answer Any Six Questions

6 x 5

- Define P<sup>H</sup> of a solution. How many grams of NaOH are required to prepare2 litres of its solution having P<sup>H</sup> 12?
- b. Explain electrolysis of aqueous solution of NaCl.
- c. Explain magnetic separation method of concentration of ores with a neat and labelled diagram.
- d. Define hardness of water. Explain cold-lime soda process of removal of hardness of water.
- e." Write a brief note on preparation and uses of Bakelite.
- f.\* Define and explain Arrhenius theory of acids and bases. What is neutralization reaction?
- g. Write down the IUPAC names/structural formulae of the following.

(i) 
$$CH_3-CH-CH = C-CH_3$$
  
 $C_2H_5$  OH

- (iv) 4-Bromo-3-methylpent-3-en-2-ol
- (v) 2-Methylbuta-1,3-diene

ŝ	Briefly explain 'Bohr's atomic model' for hydrogen atom.	0
	(b) How many grams of Ca(OH)2 are required to prepare 2.5 litres of its decinormal solution?	0
4	(a) What are lubricants? Write down the purpose of lubrication.	0
	(b) Define herbicide and fungicide with at least two examples from each.	0
M	(a) Write down the composition and uses of producer gas.	0
1	(b) Give a brief note on 'Galvanization'.	0.
6	(a) Define Faraday's 1st law of electrolysis. How many grams of silver will be deposited by the passage of	0
	50 ampere of current through silver nitrate solution of 30 minutes?	
	Define covalent bonding. Explain the formation of CH <sub>4</sub> molecule.	0
7	(a) Define and explain vulcanisation of raw rubber.	0
	(b) Distinguish between saturated and unsaturated hydrocarbons.	0